2019 Ph H1 Q16

Section: Particles and Waves

Topic: Forces on Charged Particles / Photoelectric Effect

Brief summary of the question

Photon energy: 6.40×10^{-19} J, work function: 4.20×10^{-19} J. Find maximum electron speed.

Worked solution

Photoelectric equation: K_max = hf -

K max =
$$6.40 \times 10^{-19} - 4.20 \times 10^{-19} = 2.20 \times 10^{-19}$$
 J.

$$K = \frac{1}{2} \text{ m v}^2 \quad v = \text{sqrt}(2K / m_e).$$

$$v = sqrt((2 \times 2.20 \times 10^{-19}) / (9.11 \times 10^{-31})).$$

$$v = sqrt(4.83 \times 10^{11}) = 6.95 \times 10^{5} m/s.$$

Final answer

$$C - 6.95 \times 10^5$$
 m/s.

Revision tips

- Subtract work function from photon energy to get K_max.
- Use $m_e = 9.11 \times 10^{-31} \text{ kg}$.
- Speed here is non-relativistic (well below 0.1c).