Ultraviolet radiation of frequency 7.70×10^{14} Hz is incident on the surface of a metal. Photoelectrons are emitted from the surface of the metal.

The maximum kinetic energy of an emitted photoelectron is 2.67×10^{-19} J.

The work function of the metal is

$$A \qquad 1 \cdot 07 \times 10^{-19} \, J$$

 $2.44 \times 10^{-19} \, \text{J}$

$$imes$$
 10⁻¹⁹ J

$$2.44 \times 10^{-19} \text{ J}$$

 $2.67 \times 10^{-19} \text{ J}$

F 7.78×10^{-19} L

D
$$5.11 \times 10^{-19} \text{ J}$$