

2024 Higher Chemistry Paper 2 - Q10

Section: Chemistry in Society

Topic: Chemical Analysis (Hydrogen NMR)

Question summary (Q10):

Hydrogen NMR spectra provide information about hydrogens in different chemical environments.

(a) Using the data booklet, state the chemical shift range for hydrogens in the environment:

$\text{-C}\equiv\text{C-H}$ (alkyne proton).

Worked Solution:

- In ^1H NMR, different environments give characteristic chemical shift ranges (δ , ppm).
- Alkyne hydrogens ($\text{-C}\equiv\text{C-H}$) resonate at relatively high field values compared with alkenes.
- From the data booklet, $\text{-C}\equiv\text{C-H}$ protons appear in the range: 2.5 – 3.0 ppm.

Final Answer:

Chemical shift range for $\text{-C}\equiv\text{C-H}$ hydrogens: 2.5 – 3.0 ppm

Revision Tips:

- Chemical shift values depend on electron density around the proton (deshielding).
- Order (approximate ranges, ppm):
Alkane 0.5–1.5 Alkyne 2.5–3.0 Alkene 4.5–6.5 Aromatic 6.0–9.0
Aldehyde 9.4–10.0 Carboxylic OH 10–13
- Remember: number of peaks = number of distinct environments; peak height/area = number of H in each environment.