2

- Titanium is a metal which is expensive to extract and purify.It can be extracted from an ore in a two-step process called the Kroll process.
  - (a) The first step involves heating the ore with chlorine and carbon to convert titanium dioxide,  $TiO_2$ , to titanium chloride,  $TiCl_4$ .

$$TiO_2$$
 +  $2Cl_2$  +  $C$   $\rightarrow$   $TiCl_4$  +  $CO_2$   $GFM = 79.9 g$   $GFM = 71.0 g$ 

The largest reactors can process 1600 kg of titanium dioxide per day.

(i) Calculate the mass of chlorine, in kg, required to react with 1600 kg of titanium dioxide, TiO<sub>2</sub>.

(ii) Titanium chloride is a liquid at room temperature.Suggest the type of bonding and structure present in titanium chloride.1

\* X 8 1 3 7 6 0 1 3 0 \*

