In the reaction between potassium sulfite and iodine dissolved in potassium iodide, the species most likely to act as the oxidising agent is

species most likely to act as the oxidising agent is 
$$A \qquad I^-(aq)$$

 $I_2(aq)$ 

 $SO_3^{2-}(aq)$  $SO_4^{2-}(aq)$