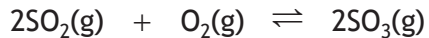


20. The equation represents a mixture at equilibrium.



Which line in the table is true for the mixture after a further two hours of reaction under the same conditions?

	Rate of forward reaction	Rate of back reaction
A	decreases	decreases
B	increases	increases
C	unchanged	decreases
D	unchanged	unchanged