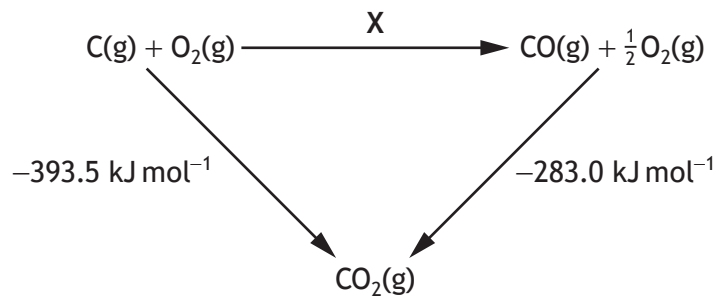


18. Consider the reaction pathway shown.



According to Hess's law, the enthalpy change, in kJ mol^{-1} , for reaction X is

- A -676.5
- B -110.5
- C $+110.5$
- D $+676.5$

