## 2024 Higher Chemistry Paper 1 - Q8

Section: Chemical Changes & Structure Topic: Rates of Reaction (Relative rate)

## Question summary (Q8):

The relative rate of a reaction which reached completion in 3 minutes 20 seconds is:

 $A \ 0.005 \ s^{-1}$   $B \ 0.313 \ s^{-1}$   $C \ 0.005 \ min^{-1}$   $D \ 0.313 \ min^{-1}$ 

## Worked Solution:

- For completion-time questions, relative rate = 1 / time.
- Convert 3 minutes 20 seconds to seconds:  $3 \times 60 + 20 = 200$  s.
- Relative rate =  $1/200 \text{ s} = 0.005 \text{ s}^{-1}$ .
- In minutes: 3 min 20 s  $\approx$  3.33 min, so 1 / 3.33  $\approx$  0.300 min<sup>-1</sup> (not in options).

Final Answer:  $A - 0.005 \text{ s}^{-1}$ 

## **Revision Tips:**

- Always put time into the unit used in the options before taking the reciprocal.
- Relative rate is a quick-comparison tool; units matter ( $s^{-1}$  vs min<sup>-1</sup>).
- If options mix units, convert your time accordingly and check for common traps like using minutes when seconds are required.